

UPSC

SYNTHESIS, CRYSTAL STRUCTURE AND HIRSHFELD SURFACE ANALYSIS OF IONIC **ADDUCT OF 2,4-DIAMINO-6-PHENYL-1,3,5-TRIAZINE WITH MALONIC ACID** Nicoleta CRACIUN^{a,b}, Elena MELNIC^{a,b}, Diana CHIȘCA^{a,b}



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> 2,4-Diamino-6-phenyl-1,3,5-triazine (dpt) is a popular molecule widely used both in coordination and supramolecular chemistry due to the availability of nitrogen-rich organic triazine core acting as a good donor and acceptor in supramolecular synthesis to design materials with desired physicochemical properties. Co-crystallization of dpt with various organic and inorganic acids resulted in multidimensional supramolecular architectures [1, 2]





a/Å	8.4392(13)
b/Å	5.1001(6)
c/Å	29.207(5)
α	90
β	92.719(14)
γ	90

ANALYSIS

ENERGY EVALUATION

Hirshfeld surface analysis revealed that the $H \cdots H$ (34.5%), O···H (18.4%) and N···H (13.1%) interactions make the highest contributions to the Hirshfeld surface.



The intermolecular energy evaluation calculated according to the energy model CE-B3LYP with 6-311G (d,p) basis set revealed that the N-H···N hydrogen bonds from homosynthon are more effective, having a higher energy value (-39.4 kJ/mol) compared to the N-H···O hydrogen bonds in heterosynthons (-18.9 kJ/mol and -7.4 kJ/mol), thus confirming significance of such interactions for

CONCLUSION Synthesis and characterization of this structure of basic dpt and flexible malonic acid are reported. The X-ray studies show that this structure represent an organic salt. In the result of the Hirshfeld analysis it was observed that the N-H…N hydrogen bonds from homosynthon are more effective, having a higher energy value -39.4 kJ/mol.

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